



Animal slurry acidification: more than a solution for ammonia emissions abatement?

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Structure of the presentation

- 1) Introduction**
- 2) Slurry composition and separation**
- 3) Gaseous emissions during storage**
- 4) N, P and C dynamics after soil application**
- 5) Agronomic value**

Introduction

- ✓ **pH lowering of animal manure: logical and direct solution to minimize ammonia emissions.**



- ✓ **Additives used: nitric and sulfuric acid with liquid manure and aluminium sulfate with solid manure**

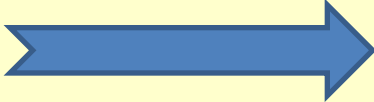
Introduction

- ✓ **Today, safe and efficient solutions are proposed to farmers for slurry acidification in barn, in slurry store or immediatly before soil application**
- ✓ **But such service is still limited to Denmark where 15% of slurry was acidified in 2013 with an expected increase to 20% in 2014**
- ✓ **More information is needed to export such technology to countries from South Europe**



Introduction

- **Strong focus of research on efficiency to decrease NH_3 but few data on other gas emissions neither on N, P and C dynamics in soil;**
- **Most studies performed in North Europe and few available information on the applicability of this technique in Mediterranean countries where pedo-climatic conditions are very different from North Europe.**

 **Provide an overview about the potential for slurry acidification application as a slurry management tool in Mediterranean countries.**

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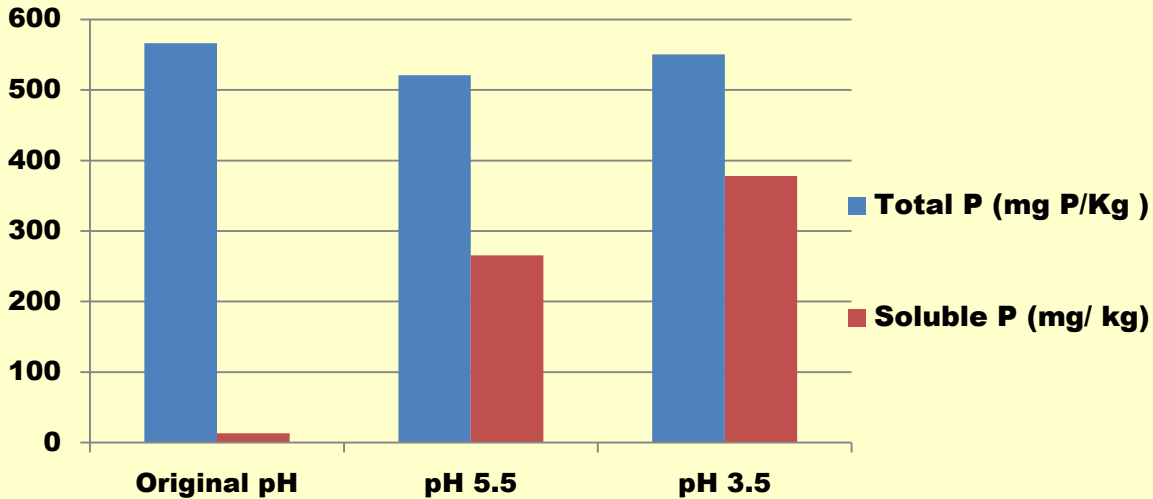
5) Agronomic value

Slurry composition and separation

	Pig slurry	Acidified pig slurry
Dry matter content (g kg⁻¹)	49.6	62.6
Total N (g kg⁻¹)	4.2	4.3
Total P (g kg⁻¹)	1.1	1.0
Total C (g kg⁻¹)	14.9	13.3
Inorganic C(g kg⁻¹)	0.8	0.0
Ca (g kg⁻¹)	2.4	2.4
Mg (g kg⁻¹)	1.1	1.0

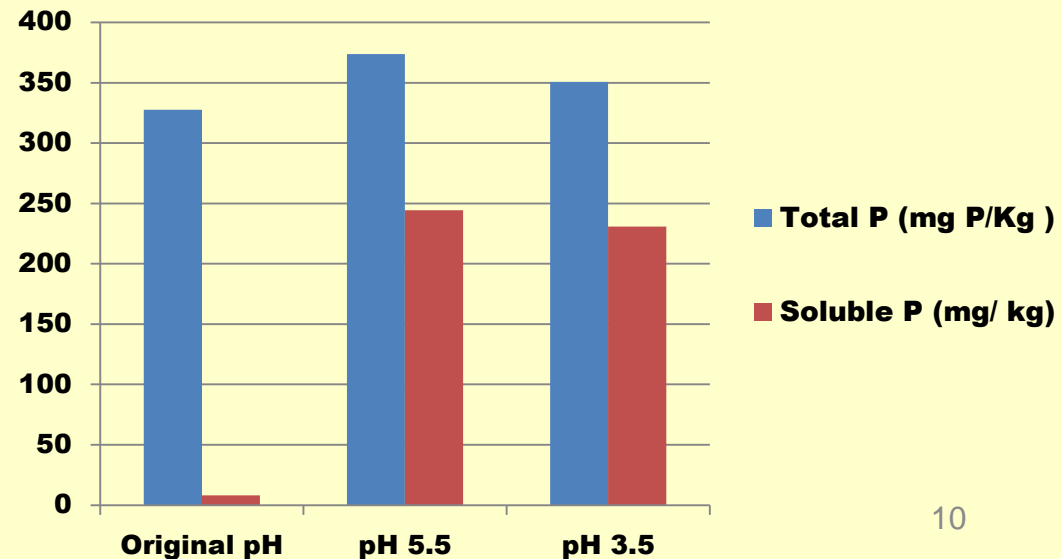
Slurry composition

Phosphorous - dairy slurry



Influence of pH target storage time

Phosphorous - pig slurry



Possible dissolution of the dominants mineral P species in manure (struvite and di-calcium phosphate)

Slurry composition and separation

	Slurry	Acidified Slurry	Liquid fraction	Acidified Liquid fraction	Solid fraction	Acidified Solid fraction
Dry matter content (g kg⁻¹)	49.6	62.6	10.2	30.9	194.8	195.5
Total N (g kg⁻¹)	4.2	4.3	2.8	2.9	10.4	9.9
Total P (g kg⁻¹)	1.1	1.0	0.04	0.69	4.7	2.1
Total C (g kg⁻¹)	14.9	13.3	2.7	1.9	72.4	67.6
Inorganic C (g kg⁻¹)	0.8	0.0	0.4	0	2.4	1.0
Ca (g kg⁻¹)	2.4	2.4	0.1	0.9	10.6	6.7
Mg (g kg⁻¹)	1.1	1.0	0.05	0.5	4.7	2.8

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Gaseous emissions during storage

Methane Emissions

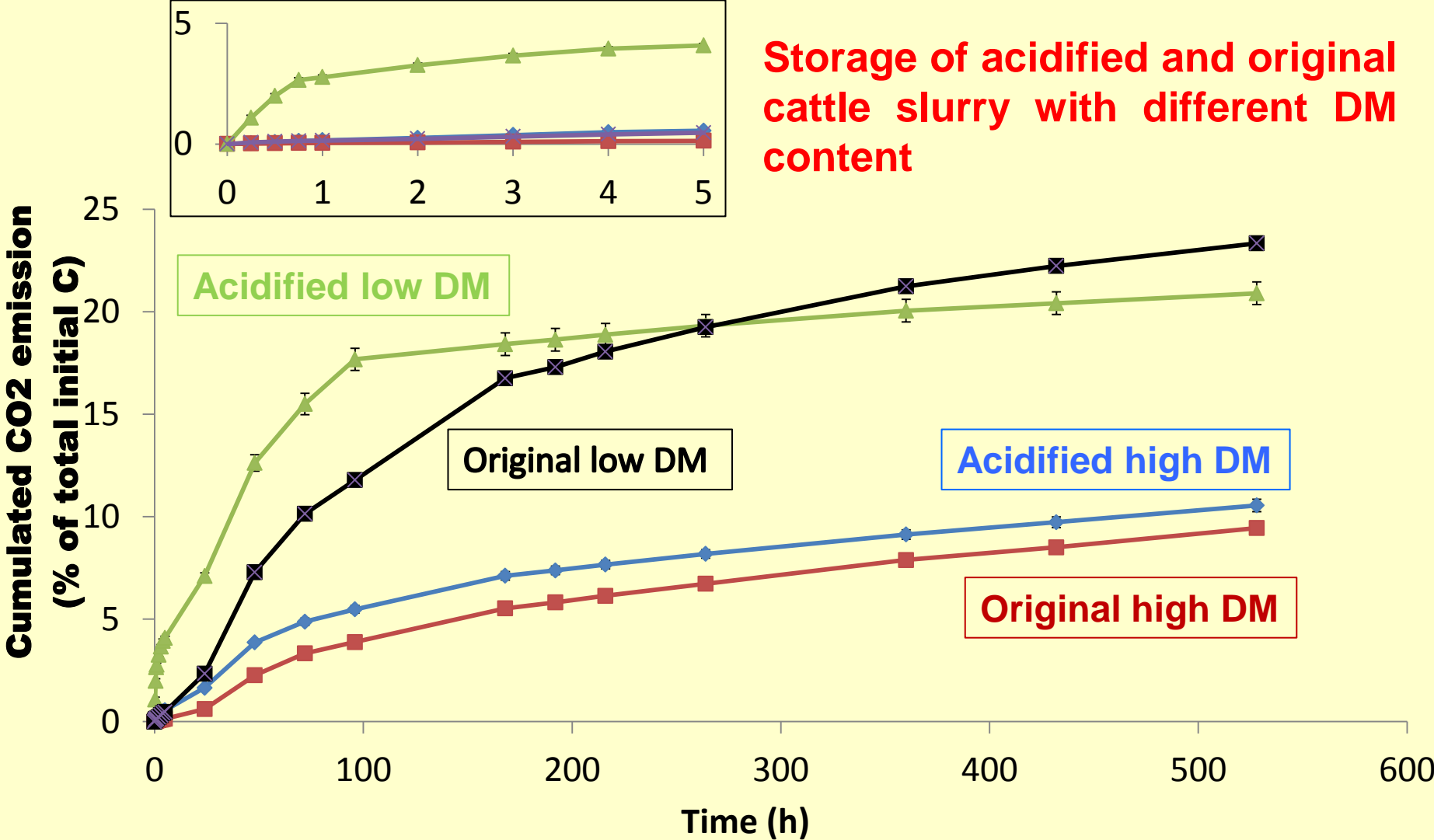
- **Slurry acidification can decrease CH₄ emissions during storage (Berg et al., 2003)**
- **Acidification efficiency to decrease CH₄ emissions depends strongly on the acid used (Berg et al., 2003; Petersen et al.2012) :**
 - **>90% with lactic acid**
 - **40-65 % with HCl**
 - **17-75% with nitric acid**
- **Below pH 5, this decrease does not depends on the target pH .**

Gaseous emissions during storage

Carbon dioxide emissions

- ✓ **CO₂ emission occurred mainly during the acidification process and can be 2-10 times higher than during subsequent storage (Fangueiro et al., 2013; Dai and Blanes Vidal, 2013).**
- ✓ **A stronger and faster decay of CO₂ emissions is observed in acidified slurry relative to non acidified during the first days of storage.**
- ✓ **Over the whole storage period, differences between acidified and non acidified slurry in terms of CO₂ emissions were generally not significant.**

Gaseous emissions during storage



Fangueiro D., Surgy S., Coutinho J., Vasconcelos E. 2012 Impact of cattle slurry acidification on carbon and nitrogen dynamics during storage and after soil incorporation. *Journal of Plant Nutrition and Soil Science*, In press

Gaseous emissions during storage

Hydrogen sulfide emissions

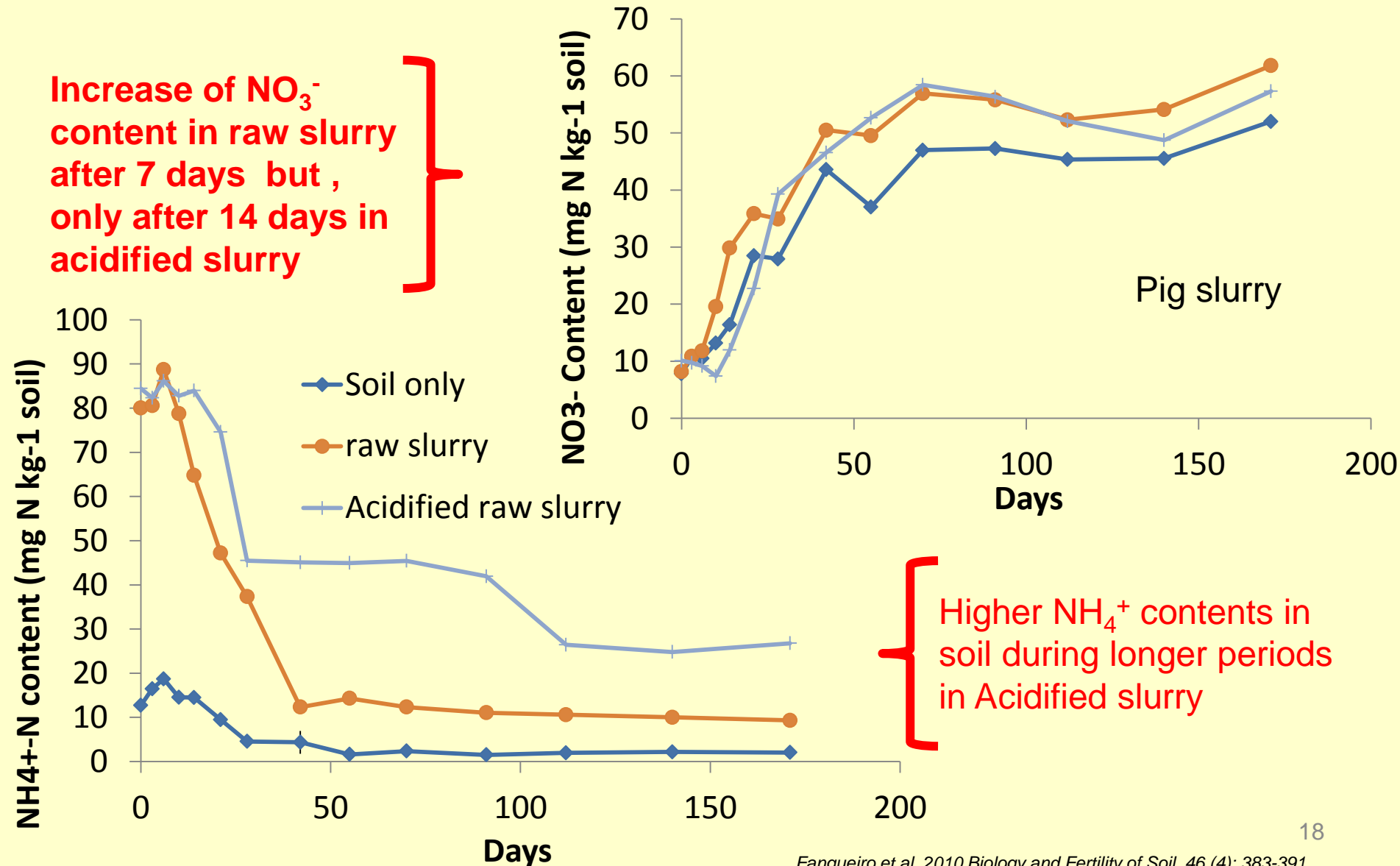
- ✓ **As occurred with CO₂ emissions, a strong burst of H₂S emissions may happened during the acidification process followed by a strong decrease over the first days of subsequent storage.**
- ✓ **BUT acidification has no significant effect on H₂S emissions over the whole storage period (Dai and Blanes-Vidal (2012))**
- ✓ **More than the pH effect, the slurry mixing that was performed in all treatments strongly influence H₂S emissions.**

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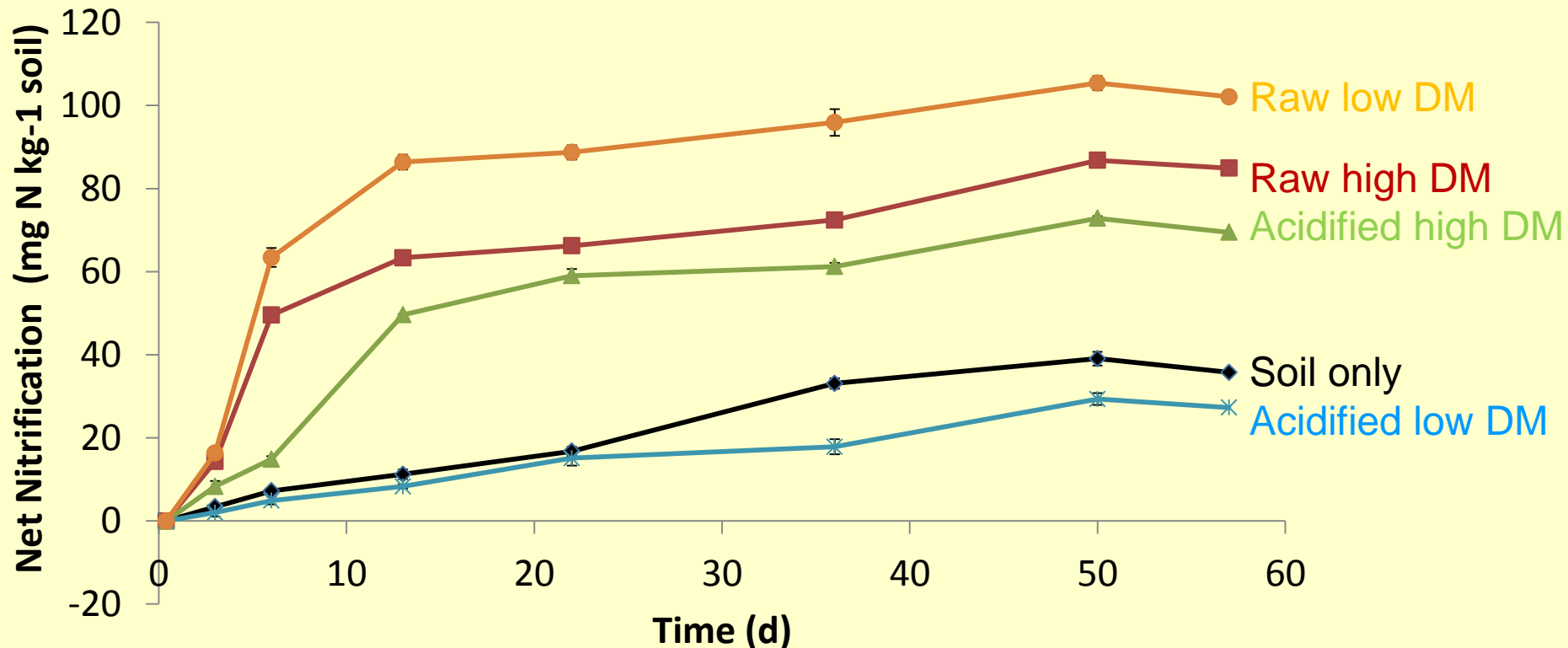
Influence of slurry acidification on N mineralization and nitrification

Increase of NO_3^- content in raw slurry after 7 days but, only after 14 days in acidified slurry



Influence of slurry acidification on N mineralization and nitrification

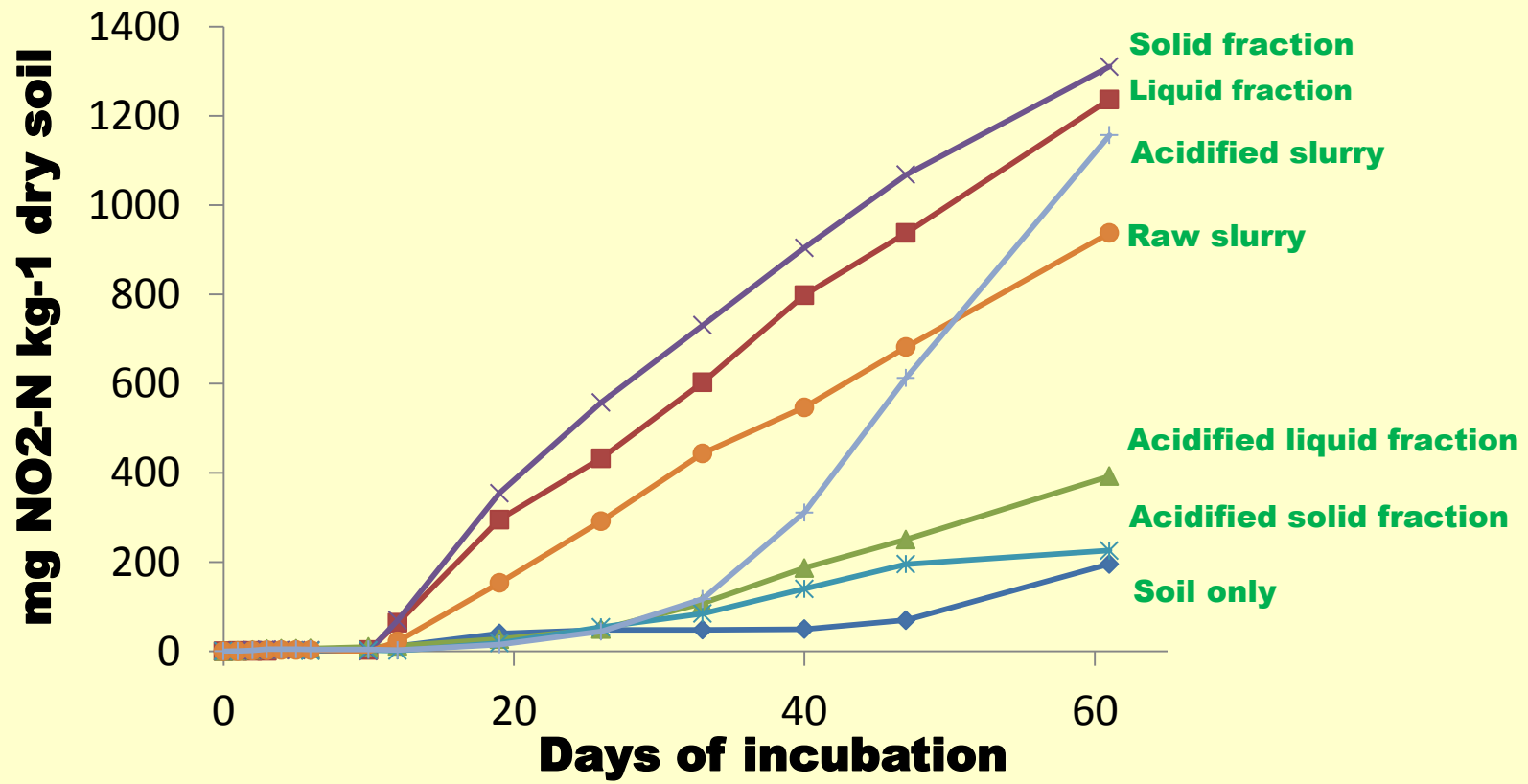
Delay and decrease of nitrification with acidification: effect more intense in low DM slurry



Soil application of acidified and non acidified cattle slurry with different DM content

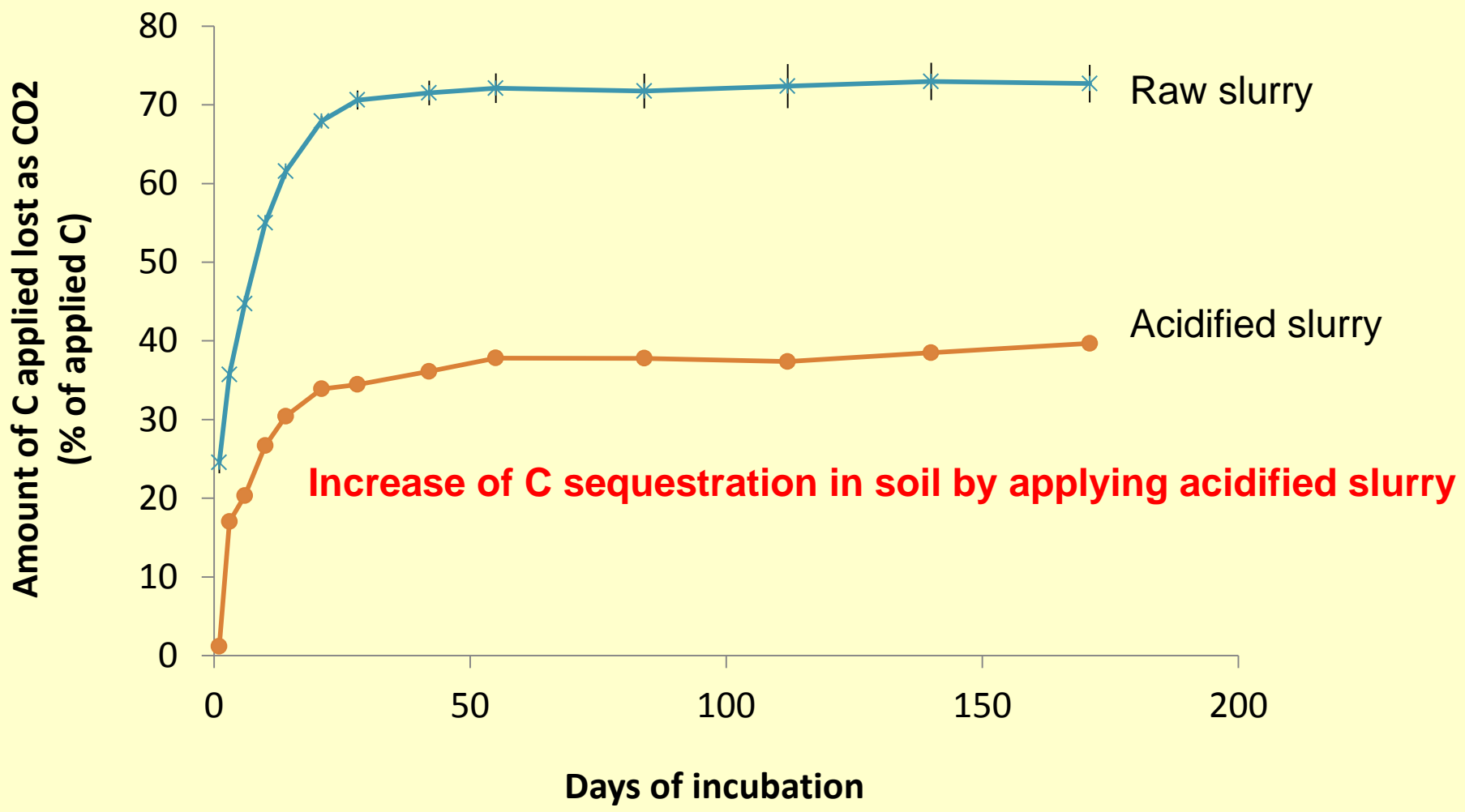
Influence of slurry acidification on N₂O after soil application – pig slurry

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Fangueiro D., Ribeiro H., Coutinho J., Cardenas L., Trindade H., Cunha-Queda C., Vasconcelos E., Cabral F. 2010 Nitrogen mineralization and CO₂ and N₂O emissions in a sandy soil amended with original or acidified pig slurries or with the relative fractions. *Biology and Fertility of Soil*, 46 (4): 383-391.

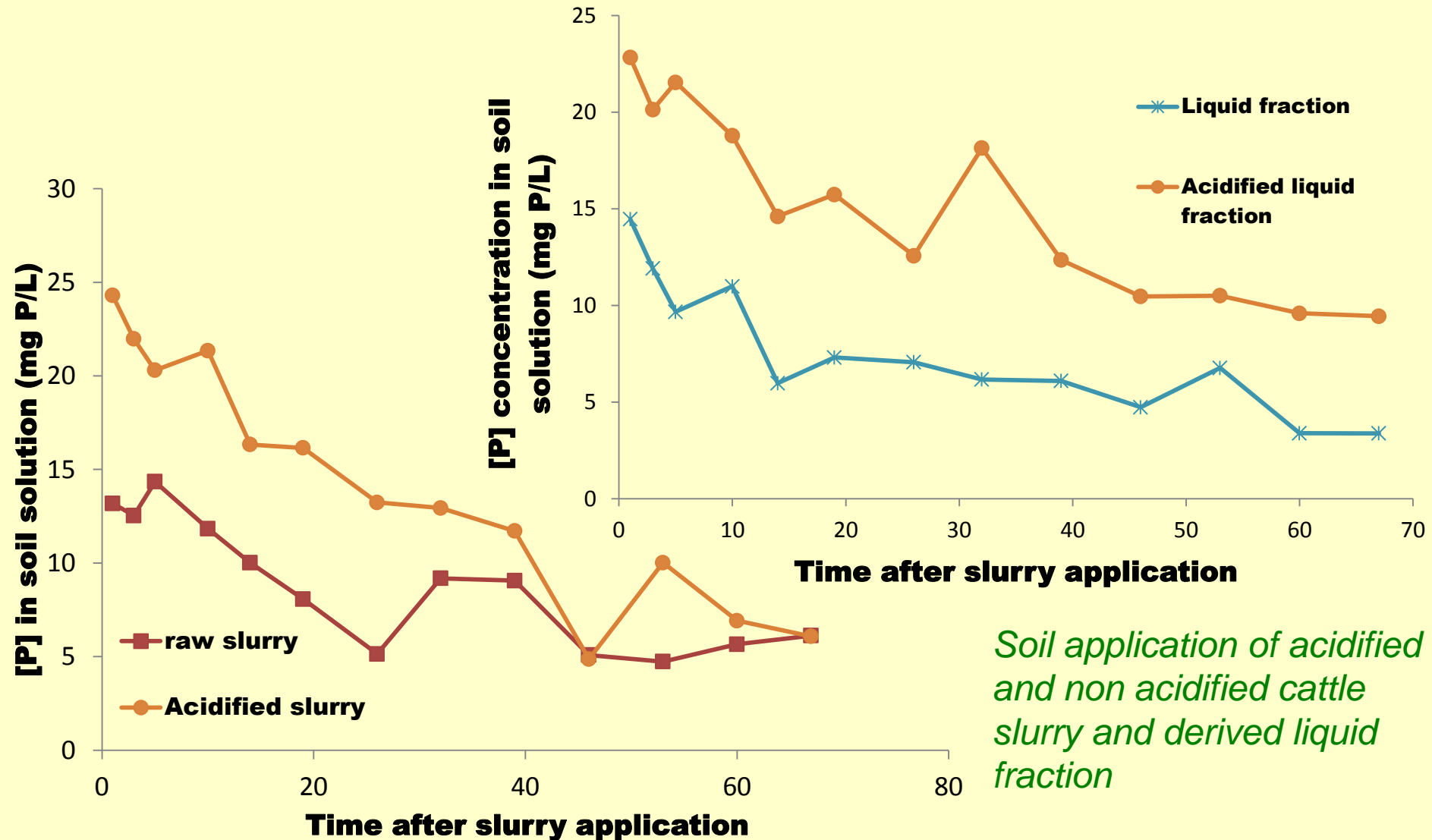
Influence of slurry acidification on CO₂ emissions after soil application – pig slurry



Fangueiro D., Ribeiro H., Coutinho J., Cardenas L., Trindade H., Cunha-Queda C., Vasconcelos E., Cabral F. 2010 Nitrogen mineralization and CO₂ and N₂O emissions in a sandy soil amended with original or acidified pig slurries or with the relative fractions. *Biology and Fertility of Soil*, 46 (4): 383-391.

Influence of slurry acidification on P availability

Slurry acidification increases P availability for plants



Soil application of acidified and non acidified cattle slurry and derived liquid fraction

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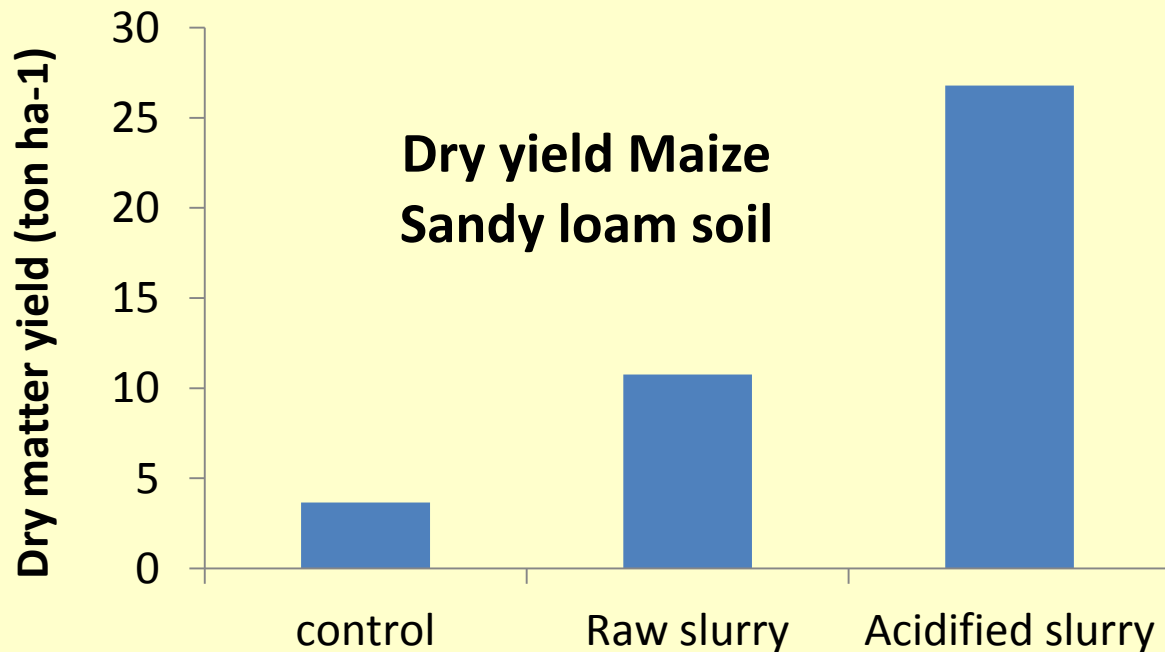
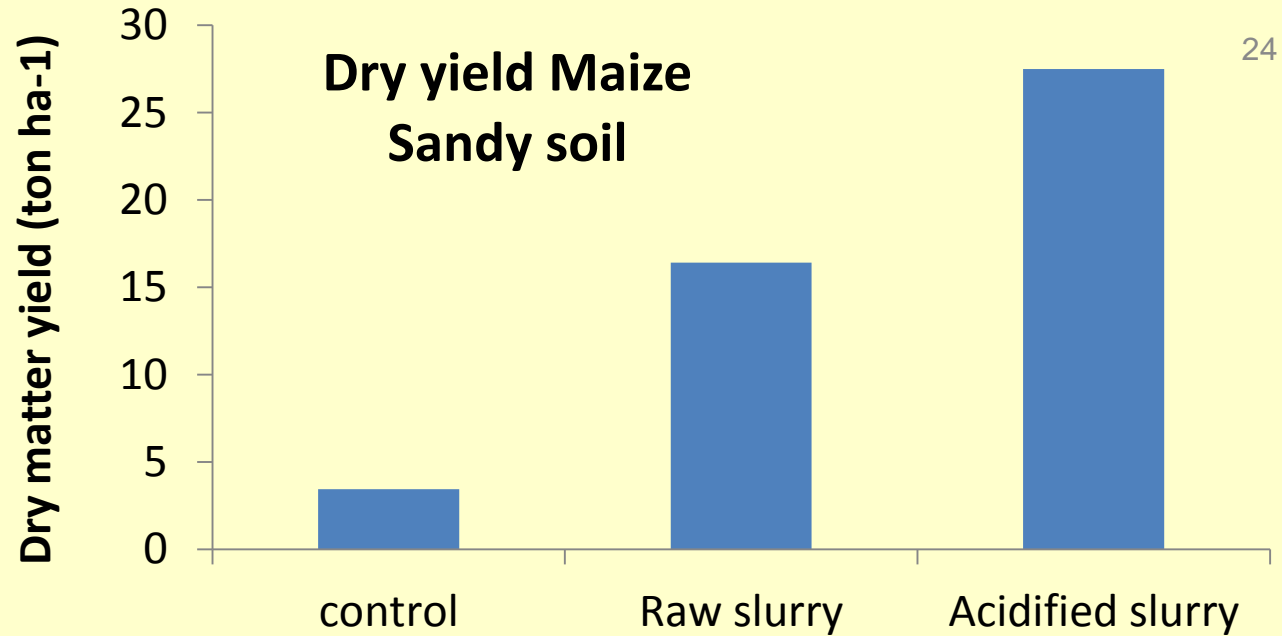
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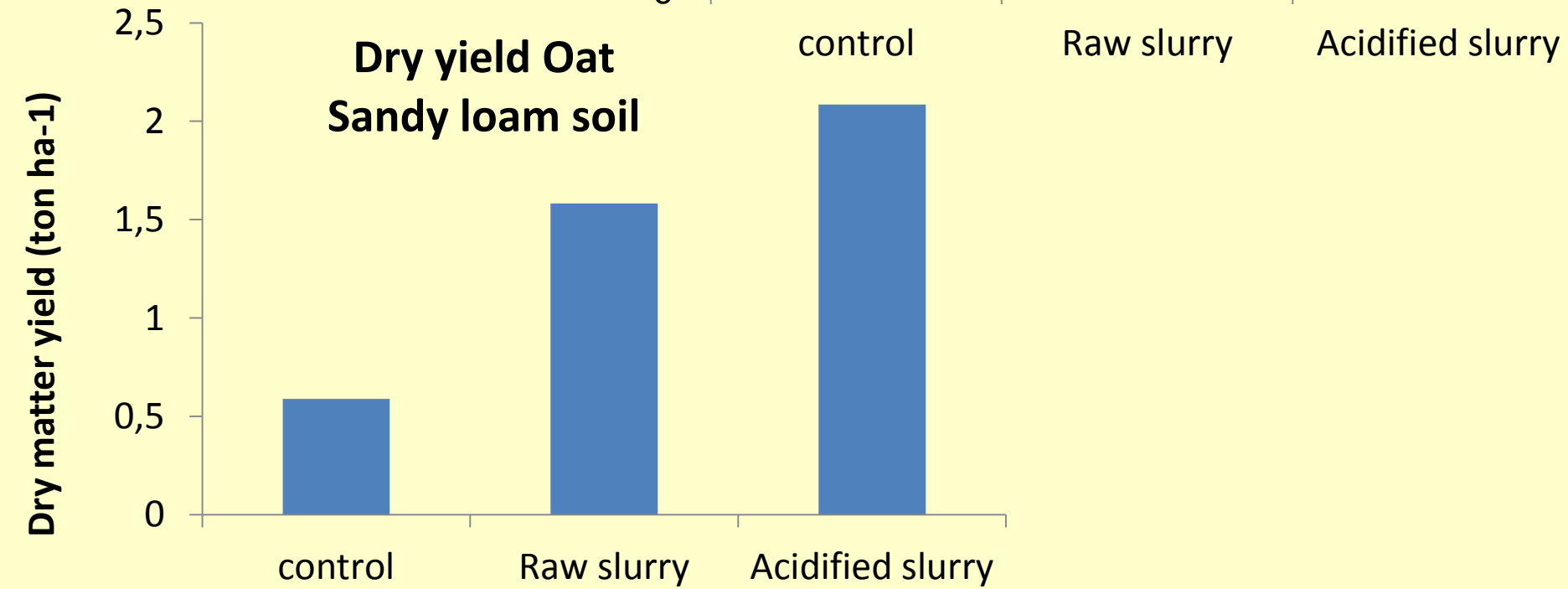
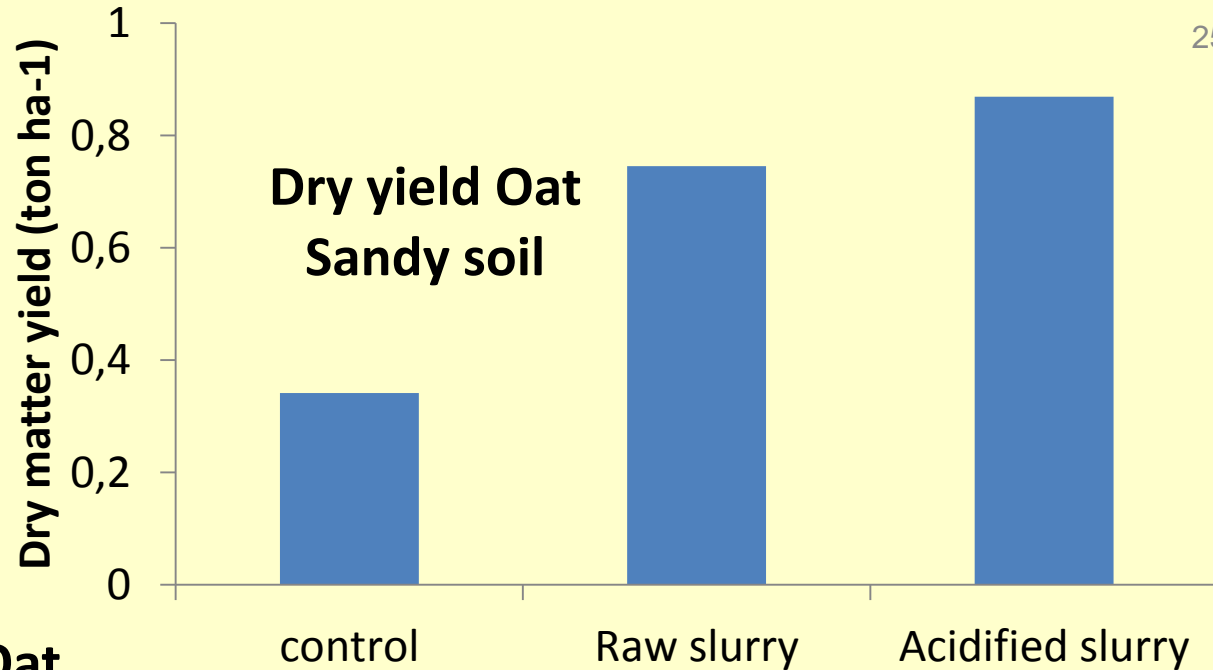
Agronomic Value

Acidified slurry may act as a starter for maize (Petersen et al., 2013)



Agronomic Value

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Conclusions

Slurry composition	Soluble P	↗
	Inorganic C	↘
Storage	CO₂	→ (initial burst)
	CH₄	↘
	H₂S	→ (initial burst)
Soil application	Nitrification	↘ (delay)
	N₂O	→ ↘
	CO₂	↘
	P availability	↗
	Crop yields	↗

**YES, slurry acidification can be used as
a slurry management tool in
Mediterranean countries**

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I. Regueiro,
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J. Coutinho



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Thank you for you attention



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